Iminophosphorane-Mediated Synthesis of 1-Substituted-β-Carbolines: Investigative Studies on the Preparation of Alkaloids Lavendamycin and Eudistomins Framework.

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Abstract.- Aza-Wittig reaction of iminophosphorane 3, derived from ethyl α -azido- β -(3-indolyl) propenoate and tri - phenylphosphine, with 2-formylpyrrole and 2-formylquinolines leads to Euclistomins A and M and Lavendamycin derivatives respectively.

Alkaloids containing a β -carboline ring system substituted at position 1 by a heteroaryl ring represent an important class of compounds which display interesting biological activities. Lavendamycin 1, isolated from *Streptomyces Lavendulae* and structurally and biogenetically related to Streptonigrin, displays an important antimicrobial and antitumor activity against P-388 and L-1210¹, consequently it has been the focus of synthetic



efforts since its initial structural identification. Synthetic approaches towards 1 involve either Bischler-Napieralski reaction between β methyltryptophan and quinaldic acid derivatives² or Pictet-Spengler cyclization reaction between β -methyltryptophan and 8-benzyloxy-2-formylquinoline³ to construct the ring C. A recent total synthesis of 1 involves formation of the B ring by Friedlander cyclization between an o-aminobenzaldehyde possesing suitable functionality and an 1acetyl- β -carboline derivative, available by regioselective inverse electron demand [4+2] cycloaddition followed by palladium (0)mediated β -carboline synthesis⁴. The alkaloids Eudistomins A and M 2 which contain the rare 1-pyrrolinyl- β -carboline ring system, have been isolated⁵ from the active antiviral Caribbean tunicate *Eudistoma Olivaceum*. The Eudistomin M has been prepared⁶ in five-steps from 6-methoxy-1,2,3,4-tetrahydro- β -carboline-1-carboxylic acid. In the course of our studies directed towards the synthesis of fused

heterocycles based on heterocyclization reaction of unsaturated heterocumulenes, we have developed the so-called tandem aza-Wittig/electrocyclization strategy for the synthesis of pyridines and isoquinoline derivatives⁷. We now report a slight modification of this strategy, undertaken in order to prepare some 1-substituted β -carboline derivatives of valuable biological interest. Our approach is centered on the aza Wittig reaction of the iminophosphorane3 derived from ethyl α -azido- β -(3-indolyl)propenoate with aldehydes to give 2azahexatrienes which undergo cyclization to give 1-substituted- β -carbolines.

The starting iminophosphorane 3, easily prepared from 3-formyl-1-methyl indole by sequential treatment with ethyl azidoacetate and triphenylphosphine⁸, reacts with aldehydes in toluene at 160°C in a sealed tube to give 1-substituted- β -carbolines 4 in 70-88% yields. The conversion $3\rightarrow 4$ involves initial aza-Wittig reaction to give a 2-azahexatriene which subsequently undergoes electrocyclic ring-closure⁹. Further dehydrogenation under the reaction conditions leads to the β -carbolines 4 (Scheme 1).



In this way, iminophosphorane 3 also reacts with 2-formylpyrrole to give 5 in 70% yield¹⁰ thus completing the assemblage of the carbon skeleton of Eudistomins A and M.

Reaction between iminophosphorane 3 and 2-formylquinoline derivatives is carried out under the same conditions. It is gratifying to note that the reaction goes smoothly to offer 6a and 6b, which contain the Lavendamycin ABCDE framework (Scheme 2).



The presence in **6b** of a benzyloxy group at C-8 is exploited to introduce bromine at positions C-5 and C-7 which eventually³ rendered the aminoquinone ring A.

Finally, iminophosphorane **3** reacts with pyruvaldehyde to give the key intermediate 1-acetyl- β -carboline **7** in 55% yield¹¹, which undergoes aza-Wittig reaction followed by aldol condensation with concomitant hydrolisis of the ester group by the action of the iminophosphorane derived from o-azidobenzaldehyde¹² in the presence of sodium ethoxide to give **6c** in 75 % yield. Friedlander condensation between **7** and 2-amino-3-benzyloxy-4-bromo benzaldehyde in the presence of benzyltrimethylammonium hydroxide yields **6d** in 61% yields¹³ (Scheme 3).



Reagents and conditions: a) pyruvaldehyde, toluene, sealed tube, 160° C; b) N-(o-formylphenyl)triphenyliminophosphorane, EtOH-NaOEt, -10° C, 30 min \rightarrow reflux, 18h; c) 2-amino-3-benzyloxy-4-bromobenzaldehyde, THF, benzyltrimethylammonium hydroxide, rt, 18h.

Scheme 3

In conclusion the results reported here show that the tandem aza-Wittig/electrocyclic ring-closure strategy affords a new and versatile entry to a variety of 1-substituted- β -carbolines. This method, which is simple and direct, leads from the readily available iminophosphorane 3 to Eudistomin and Lavendamycin derivatives¹⁴. Further studies are underway in our laboratory which are aimed at the application of this methodology to the formal total synthesis of Eudistomins A and M and Lavendamycin,

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References and Notes

- Balitz, D. M.; Bush, J. A.; Bradner, W. T.; O'Herron, F.A.; Nettleton, D. E. J. Antibiot. 1982, 35, 259; Erickson, W. R.; Gould, S. J. J. Am. Chem. Soc. 1985, 107, 5831.
- 2. Kende, A. S.; Ebetino, F. H. Tetrahedron Lett. 1984, 923; Rao, A. V. R.; Chavan, S.P.; Sivadasan, L.

Tetrahedron 1986, 42, 5065.

- Hibino, S.; Okazaki, M.; Sato, K.; Morita, I.; Ichikawa, M. Heterocycles 1983, 20, 1957; Hibino, S.; Okazaki, M.; Ichikawa, M.; Sato, K.; Ishizu, T. Heterocycles 1985, 23, 261.
- Boger, D.L.; Panek, J. S. Tetrahedron Lett. 1984, 3175; Boger, D. L.; Duff, S. R.; Panek, J. S.; Yasuda, M., J. Org. Chem. 1985, 50, 5782; Boger, D. L.; Duff, S. R.; Panek, J. S.; Yasuda, M. J. Org. Chem. 1985, 50, 5790.
- 5. Kobayashi, J.; Harbour, G. C.; Gilmore, J.; Rinehart Jr, K. L. J. Am. Chem. Soc. 1984, 106, 1526.
- Rinehart Jr, K. L.; Kobayashi, J.; Harbour, G. C.; Gilmore, J.; Mascal, M.; Holt, T. G.; Shield, L. S.; Lafargue, F. J. Am. Chem. Soc. 1987, 109, 3378.
- 7. Molina, P.; Fresneda, P. M.; Almendros, P. Tetrahedron 1991, 47, 4175.
- 8. Molina, P.; Fresneda, P. M. J. Chem. Soc. Perkin Trans 1 1988, 1819.
- Molina, P.; Fresneda, P. M.; Alarcon, P. Tetrahedron Lett. 1988, 379; Barluenga, J.; Ferrero, M.; Palacios, F. J. Chem. Soc. Perkin Trans 1 1990, 2193.
- 10. Compound 5. ¹H n.m.r. (200 MHz, CDCl₃) δ 1.38 (t, 3H, ³J=7.1Hz), 3.82 (s, 3H, *CH*₃-N), 4.37 (q, 2H, ³J=7.1Hz), 6.31 (q, 1H, ³J= 3.2Hz, H-4'), 6.52 (m,1H, H-3'), 6.94 (m, 1H, H-5'), 7.35 (t, 1H, ³J=7.4, H-6), 7.47 (d, 1H, ³J=8.3Hz, H-8), 7.63 (td, 1H, ³J=7.4Hz, ⁴J=0.6Hz, H-7), 8.11 (d, 1H, ³J=7.8, H-5), 8.48 (s, 1H, H-4), 10.32 (s br, 1H, NH); ¹³C n.m.r. (50 MHz, CDCl₃) δ 14.33 (CH₃), 33.11 (*CH*₃-N), 61.28 (CH₂), 109.06 (C-4'), 110.40(C-8), 111.89 (C-3'), 115.17 (C-5'), 119.78 (C-4), 120.79 (C-5), 121.42 (C-6), 121.56 (C-4a), 128.24 (C-2'), 128.78 (C-7), 130.51 (C-4b), 135.70 (C-9a), 136.50 (C-8a), 136.76 (C-3), 143.68 (C-1), 165.86 (CO); m/z (%) 319 (M+, 26).
- 11. Compound 7. ¹H n.m.r. (200 MHz, CDCl₃) δ 1.51 (t, 3H, ³J=7.1Hz), 3.00 (s, 3H, *CH*₃-CO), 3.97 (s, 3H, *CH*₃N), 4.53 (q, 2H, ³J=7.1Hz), 7.39 (t, 1H, ³J=7.4Hz, H-6), 7.54 (d, 1H, ³J=8.2Hz, H-8), 7.68 (t, 1H, ³J=7.4Hz, H-7), 8.20 (d, 1H, ³J=7.7Hz, H-5), 8.96 (s, 1H, H-4); ¹³C n.m.r. (50 MHz, CDCl₃) δ 14.41 (CH₃), 28.34 (*CH*₃-CO), 34.13 (*CH*₃-N), 61.61 (CH₂), 110.44 (C-8), 119.77 (C-4), 120.98 (C-4a), 121.23 (C-6), 121.48 (C-5), 129.47 (C-7), 132.26 (C-4b), 135.87 (C-9a), 136.06 (C-8a), 139.26 (C-3), 143.62 (C-1), 165.45 (*COOEt*), 201.30 (*CO*-CH₃); m/z (%) 269 (M⁺, 61).
- Luheshi, A. B. N.; Salem, S. M.; Smalley, R. K.; Kennewell, P. D.; Westwood, R. Tetrahedron Lett. 1990, 31, 6561; Molina P.; Arques A.; Cartagena, I.; Obón, R. Tetrahedron Lett. 1991, 2521.
- 13. Compound **6d**. ¹H n.m.r. (200 MHz, DMSO-d₆) δ 3.74 (s, 3H, *CH*₃N), 5.17 (s, 2H, Ph*CH*₂O), 6.90 (d, 2H, ³J=7.6Hz, 2xH_o), 7.10 (t, 1H, ³J=7.3Hz, H_p), 7.31 (d, 2H, ³J=7.6Hz, 2xH_m), 7.40-7.55 (m, 1H, H-6), 7.65-7.76 (m, 2H, H-7 and H-8), 7.80-7.90 (m, 2H, H-6' and H-5'), 8.49 (d, 1H, ³J=8.6Hz, H-3'), 8.56 (d, 1H, ³J=7.8Hz, H-5), 8.70 (d, 1H, ³J=8.6Hz, H-4'), 9.10 (s, 1H, H-4). ¹³C n.m.r. (50 MHz, DMSO-d₆) δ 34.69 (*CH*₃N), 75.90 (Ph*CH*₂O), 111.11 (C-8), 117.03 (C-7'), 117.43 (C-4), 120.83 (C-4a), 121.02 (C-6), 122.23 (C-5), 123.49 (C-3'), 125.10 (C-5'), 128.06 (C_p and C_o), 128.42 (C-4'a), 128.62 (C_m), 129.39 (C-7), 131.05 (C-6 and C_{4b}), 136.57(C-9a), 136.92 (C-8a), 137.05 (C-8'a), 137.57 (C-4'), 141.43 (C-1), 143.44 (C-3), 151.88 (C-8'), 156.72 (C-2')166.60 (CO); m/z (%) 538 (M⁺, 5).
- 14. Satisfactory ¹H, ¹³C n.m.r. (values assigned by decoupling methods and 2D ¹H-¹³C correlation techniques), mass spectra and elemental analyses were obtained for all new compounds.